## Curriculum map

Department of Chemistry and Biochemistry/Chemistry Major
Learning outcomes (LOs): Having completed a major in Chemistry, a student will be able to:

1. Master a broad set of chemical concepts concerning the fundamentals in the basic areas of the discipline (organic, inorganic, analytical, physical and biochemistry). Students will demonstrate an understanding of structure, chemical properties, and reactions of chemicals and biomolecules.
2. Demonstrate a firm foundation in the conceptual, quantitative, and computational thinking that underlies the theories and models that form the basis for reasoning about molecular systems. Students should be able to connect this theoretical understanding to the experimental methods used to test those theories and models.
3. Demonstrate excellent critical thinking and problem solving abilities. S/he will be able to integrate chemical concepts and ideas learned in lecture courses with skills learned in laboratories to formulate hypotheses, propose and perform experiments, collect data, compile and interpret results and draw reasonable and logical conclusions. In addition, graduates will be able to rationally estimate the solution to a problem, apply appropriate techniques to arrive at a solution, test the correctness of the solution, and interpret their results.
4. Employ critical thinking and the scientific method to design, carry out, record, analyze and communicate the results of chemical/biochemical experiments. This includes the ability to identify or create an appropriate model, formulate a hypothesis, choose an appropriate set of tools and techniques, and design an experiment that tests the hypothesis and analyze the results from that experiment drawing sound scientific conclusions from the results obtained...
5. Develop the interpersonal skills to function cooperatively in a team setting.
6. Handle, synthesize, purify, and characterize new and existing substances. This includes knowing the proper procedures and regulations for the safe handling, use and disposal of chemicals.
7. Be proficient in the use of both classical and modern tools (e.g., instrumentation, techniques, software) for analysis of chemical systems. Demonstrate effective scientific communication skills, both orally and in writing, to a range of audience levels and for a variety of purposes.
8. Understand how scientific information is shared between peers in modern science, including responsible conduct for acknowledging prior and current contributions.
9. Demonstrate an awareness of the benefits and impacts of chemistry related to the environment, society, and other disciplines outside the scientific community. Be prepared to contribute solutions to society's challenges at the intersection of science and society.
10. Successfully pursue their career objectives in advanced education in professional and/or graduate schools, in a scientific career in government or industry, in a teaching career in the school systems, or in a related career following graduation.
11. Understand and apply ethics and values to all professional activities

Library-In this context, they must be able to locate, identify and critically evaluate the chemical/biochemical literature.

Key: I = introduces outcome; $D=$ develops outcome; $A=$ assesses mastery of outcome

|  | Titildescription | Foundational Knowledge/Theory |  |  |  |  |  |  |  |  | Performance/Skills Based |  |  |  |  |  | Affective |  |  |
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| Course(s) |  | LO 1 | Analytical | Biochemistry | Inorganic | Organic | Physical | Related Fields | $\begin{gathered} \text { LO } \\ 2 \end{gathered}$ | $\begin{gathered} \text { LO } \\ 3 \end{gathered}$ | $\begin{gathered} \mathrm{LO} \\ 4 \\ \hline \end{gathered}$ | $\begin{gathered} \hline \text { LO } \\ 5 \end{gathered}$ | $\begin{gathered} \text { LO } \\ 6 \end{gathered}$ | $\begin{gathered} \mathrm{LO} \\ 7 \end{gathered}$ | $\begin{gathered} \mathrm{LO} \\ 8 \end{gathered}$ | Library | LO 9 | $\begin{array}{\|c} \text { LO } \\ 10 \\ \hline \end{array}$ | $\begin{array}{\|c\|} \hline \text { LO } \\ 11 \\ \hline \end{array}$ |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| CH 221 | General Chemistry (lecture) | 1 | 1 |  | 1 |  | 1 |  | 1 |  |  |  |  |  |  |  | 1 |  |  |
| CH 222 | General Chemistry (lecture) | 1 |  |  | 1 |  | 1 |  | 1 |  |  |  |  |  |  |  | 1 |  |  |
| CH 223 | General Chemistry (lecture) | ID | 1 |  |  |  | 1 |  | 1 |  |  |  |  |  |  |  | 1 |  |  |
| CH 224 H | Honors General Chemistry (lecture) | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |  | 1 | 1 | 1 | 1 |  | 1 |
| CH 225 H | Honors General Chemistry (lecture) | ID | 1 | 1 | 1 | 1 | 1 | 1 | 1 | D | D | 1 |  | 1 | D | D | D |  | 1 |
| CH 226 H | Honors General Chemistry (lecture) | ID | 1 | 1 | 1 | 1 | 1 | 1 | 1 | D | D | 1 |  | 1 | D | D | D |  | 1 |
| CH227 | General Chemistry Laboratory | 1 |  |  | 1 |  | 1 |  | 1 | 1 | 1 | 1 | , | 1 | 1 |  | 1 |  |  |
| CH 228 | General Chemistry Laboratory | 10 | D |  |  |  | 1 |  | 1 | D | 1 | , | 1 | 1 | 1 |  | 1 |  |  |
| CH 229 | General Chemistry Laboratory | ID | D |  |  |  | 1 |  | 1 | D | 1 | 1 | 1 | 1 | 1 |  | 1 |  |  |
| CH 237 | Advanced General Chemistry Laboratory | 1 | 1 |  | 1 |  |  |  | 1 | 1 | 1 |  | 1 | D | 1 |  |  |  |  |
| CH238 | Advanced General Chemistry Laboratory | 1 | 1 | 1 | 1 |  | 1 |  | D | D | D |  | D | D | 1 | 1 |  |  |  |
| CH 239 | Advanced General Chernistry Laboratory | 1 |  | 1 | 1 |  | 1 |  | D | D | D |  | D | DA | 1 | 1 |  | 1 |  |
| CH331 | Organic Chemistry I (lecture) | 1 |  |  |  | 1 |  |  | D | D |  |  |  |  |  |  |  |  |  |
| CH 335 | Organic Chemistry II (lecture) | ID |  | 1 |  | D |  |  | D | D |  |  |  |  |  |  |  |  |  |
| CH336 | Organic Chemistry III (lecture) | ID |  | I |  | D |  |  | D | D |  |  |  |  |  |  |  |  |  |
| CH 337 | Organic Chemistry Laboratory | 1 | 1 |  |  | 1 |  |  | 1 | 1 | 1 | D | 1 | 1 | 1 |  | D |  | 1 |
| CH 338 | Organic Chemistry Laboratory | ID | 1 |  |  | D |  |  | I | D | 1 | D | D | 1 | 1 |  | A |  | 1 |
| CH341 | Majors Track Organic Chemistry I (lecture) | 1 |  |  |  | 1 |  |  | 1 | 1 |  |  |  |  |  |  | 1 |  |  |
| CH 342 | Maiors Track Organic Chemistry I (lecture) | 0 |  |  |  | D |  |  | D | D |  |  |  |  |  |  | D |  |  |
| CH343 | Majors Track Organic Chemistry I (lecture) | 0 |  |  |  | D |  |  | D | D |  |  |  |  |  |  | D |  |  |
| CH 348 | Organic Chemistry Lab for Majors | D | D |  |  | D |  |  | D | D | D | D | D | D | D | 1 | D |  | 1 |
| CH 349 | Organic Chemistry Lab for Majors | DA | D |  |  | A |  |  | A | A | A | A | D | A | A | D | D |  | 1 |
| CH360 | Physiological Biochemistry (lecture) | 10 |  | 1D |  |  |  |  | 1 | 1 |  |  |  |  | 1 |  | 1 |  |  |
| Cit 407 | Research (laboratory) | DA-specific learning outcomes are dependent on individual project |  |  |  |  |  |  | DA | DA | DA | A DA | DA | DA | D | DA | DA | D | D |
| CH 411 | Physical Chemistry: Thermodynamics (lecture) | D |  |  |  |  | D |  | D | D | D | D |  |  |  | D | D | 0 | 0 |
| CH 412 | Physical Chemistry: Kinetics (leecture) | DA |  |  |  |  | DA |  | DA | DA |  |  |  |  |  |  | D | 0 | D |
| CH413 | Physical Chemistry: Quantum Mechanics (lecture) | DA |  |  |  |  | DA |  | DA | DA |  |  |  |  |  |  | D | D | D |
| CH 417,418,419 | Physical Chemistry Laboratory | I |  |  |  |  |  |  | ID | D | D | D | D | D | D | D | D |  |  |
| CH 429 | Instrumental Analysis. Laboratory | DA | DA |  | A |  | A | A | DA | DA | DA |  | D | DA | DA | A | D | D | D |
| Other Degree Requirements |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| PHYS 201, 202, 203 OR 251, 252, 253 | General Physics (lecture) OR Foundations of Physics (lecture) |  |  | . |  |  |  |  | 1 | 1 |  |  |  |  | 1 |  |  |  |  |
| PHYS 204, 205, 206 OR 290 | Introductory Physics Laboratory OR Foundations of Physics Laboratory |  |  |  |  |  |  |  | ID | ID |  | 1 |  |  | ID |  |  |  |  |
| MTH 251, 252, 253 | Calculus I, II, III (lecture) |  |  |  |  |  |  |  | 1 |  |  |  |  |  |  |  |  |  |  |
| MTH 256 | Introduction to Differential Equations (lecture) |  |  |  |  |  |  |  | ID |  |  |  |  |  |  |  |  |  |  |
| MTH 281 | Several-Variable Calculus I (lecture) |  |  |  |  |  |  |  | ID |  |  |  |  |  |  |  |  |  |  |
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|  |  | Foundational Knowledgertheory |  |  |  |  |  |  |  | PerformanceeSkills Based |  |  |  |  |  | Affective |  |  |
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| Course(s) | Title/description | LO1 | Analytical | Biochemistry | Inorganic | Organic | Physical | Related Fields | $\begin{array}{\|c\|c\|} \hline \mathrm{LO} & \mathrm{LO} \\ \hline \end{array}$ | $\begin{gathered} \mathrm{LO} \\ 4 \end{gathered}$ | $\begin{gathered} \text { LO } \\ 5 \end{gathered}$ |  | $\begin{gathered} L 0 \\ 7 \end{gathered}$ | $\stackrel{\mathrm{LO}}{\mathrm{BO}}$ | Library | LO9 | $\begin{array}{\|c\|} \hline \mathbf{L O} \\ 10 \end{array}$ | $\left\lvert\, \begin{array}{\|l\|l\|} \hline \mathbf{L O} \\ \hline \end{array}\right.$ |
| Advanced Electives |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| CH40 | Electrochemistry (lecturelab) | DA | A |  | D | D | A |  | A A | A | D |  | A | D | A | D |  | D |
| CH 410 | Biocheristry of Disease (lecture) | ID |  | ID |  |  |  |  | ID IDA |  |  |  | DA | D | DA | 10 |  |  |
| CH 410 | Structural Biochemistry (lecture) | DA |  | DA |  |  |  |  | DAIDA |  |  |  | DA | D | DA | ID |  |  |
| CH410 | Cytoskeleton (lecture) | DA |  | DA |  |  |  |  | DA IDA |  |  |  | DA | D | DA | 10 |  |  |
| CH410 | Biophysics \& Evolution (lecture) | DA |  | DA |  |  |  |  | DAIDA |  |  |  | DA | D | DA | 10 |  |  |
| CH 420 | Physical Oraanic Chemistry ( (lecture) | A |  |  |  | A |  |  | A A |  |  |  |  |  | A |  |  | D |
| CH421 | Physical Organic Chemistry Il (lecture) | A |  |  |  | A |  |  | A A |  |  |  |  |  | A |  |  | D |
| CH431 | Inorganic Chemistry (lecture) | A |  |  |  | A |  |  | A A | A |  |  |  | A | A | A |  |  |
| CH432 | Inorganic Chemistry: Bioinorganic Chemistry (lecture) | DA |  | D | A |  | D |  | A A | A |  |  | A | A | A | A |  | A |
| CH433 | Lnorganic Chemistry: Solid-state Chemistry (lecture) | A |  |  | A |  |  |  | A A | A |  |  | A | A | A | A |  | A |
| CH437 | Inorganic Chemistry Laboratory | A |  |  | A |  |  |  | A A | A | A | A | A | D | A | D | D | D |
| CH 441 | Quantum Chemistry (lecture) | DA |  |  |  |  | DA |  | da da | dA | DA |  |  |  | DA | DA | DA | DA |
| CH 442 | Quantum Chemistry (lecture) | DA |  |  |  |  | DA |  | DA DA | DA | DA |  |  |  | DA | DA | DA | DA |
| CH443 | Quantum Chemistry and Spectroscopy (lecture) | DA |  |  |  |  | DA |  | DA DA | DA | DA |  |  |  | DA | DA | DA | DA |
| CH 444 | Chemical Thermodynamics (lecture) | DA |  |  |  |  | DA |  | DA DA | DA | DA |  |  |  | DA | DA | DA | DA |
| CH 445 | Statistical Mechanics (lecture) | DA |  |  |  |  | DA | DA | DADA | DA | D |  |  |  | D | D |  |  |
| CH 446 | Chemical Kinetics (lecture) | DA |  |  |  |  | DA |  | DA DA | DA |  |  |  |  | DA |  |  |  |
| CH447 | Computational Chemistry (lecture/lab) | DA |  | D | D | D | IDA |  | IDAIDA | IDA |  |  |  |  | DA |  | D |  |
| CH 452 | Advanced Organic Chemistry: Stereochemistry and Reactions (lecture) | A |  |  |  |  |  |  | A A |  |  |  |  |  | A |  |  | D |
| CH461 | Biochemistry: Structure and Function Macromolec (lecture) | DA |  | DA |  |  |  |  | ID DA |  | DA |  | 10 | 10 | DA | ID |  |  |
| CH462 | Biochemistry: Metabolism (lecture) | DA |  | DA |  |  |  |  | DA DA |  | DA |  | D | D | DA | 10 |  |  |
| CH 463 | Biochemistry: Mechanisms of Requation (lecture) | DA |  | DA |  |  |  |  | DA DA |  | DA |  | D | D | DA | 10 |  |  |
| CH 464 | RNA Biochemistry llecture) | DA |  | DA |  |  |  |  | 10 DA |  |  |  | DA | D | DA | ID |  |  |
| CH465 | Physical Biochemistry (lecture) | DA |  | DA |  |  |  |  | DA DA |  |  |  | DA | D | DA | ID |  |  |
| CH467 | Biochemistry Laboratory | DA |  | DA |  |  |  |  | DA DA | $\bigcirc$ | DA |  | IDA | D | DA | 10 | DA | 10 |
| GEOL 471 | Thermodynamic Geochemistry (lecure) |  |  |  | DA |  |  | IDA | DA |  |  |  |  |  |  |  |  |  |
| GEOL 472 | Aqueous-Mineral-Gas Equilibria (lecture) |  |  |  | DA |  |  | IDA | DA |  |  |  |  |  |  |  |  |  |
| GEOL473 | Isotope Geochemistry (lecture) |  |  |  | DA |  |  | IDA | DA |  |  |  |  |  |  |  |  |  |
| PHYS 412,413 | Mechanics, Electricity, and Magnetism (lecture) |  |  |  | DA |  |  | DA | DA |  |  |  |  |  |  |  |  |  |
| - PHYS 414.415 | Quantum Physics (lecture) |  |  |  | DA |  |  | DA | DA |  |  |  |  |  |  |  |  |  |

## Learning outcomes explanations

Department of Chemistry and Biochemistry/Chemistry Major
Majors in chemistry and biochemistry provide training for students planning careers in the chemical and biological sciences and also for those in biology, health related disciplines, earth sciences, secondary education, business, journalism and
law. Approximately one quarter of the UO undergraduate population will take a course in the Department of Chemistry and Biochemistry. The Department's curriculum is designed to satisfy the diverse needs of all these students.

Chemistry and biochemistry graduates complete an integrated, rigorous program that includes foundational course work in chemistry and biochemistry and additional course work in related fields. The ACS-certified degree further emphasizes laboratory experience and the development of professional skills. Undergraduate research and other educational activities outside the traditional classroom are essential components of these majors. Undergraduate majors also benefit from taking graduate courses in synthetic, physical, materials, computational chemistry, biochemistry molecular biology and modern instrumental techniques.

Graduates of our program will have a robust set of fundamental competencies that are knowledge-based, performance/skills-based, and affective.

## Foundational knowledge/theory

All our graduates will be able to:

- Master a broad set of chemical concepts concerning the fundamentals in the basic areas of the discipline (organic, inorganic, analytical, physical and biochemistry). Students will demonstrate an understanding of structure, chemical properties, and reactions of chemicals and biomolecules.
- Demonstrate a firm foundation in the conceptual, quantitative, and computational thinking that underlies the theories and models that form the basis for reasoning about molecular systems. Students should be able to connect this theoretical understanding to the experimental methods used to test those theories and models.
- Demonstrate excellent critical thinking and problem solving abilities. S/he will be able to integrate chemical concepts and ideas learned in lecture courses with skills learned in laboratories to formulate hypotheses, propose and perform experiments, collect data, compile and interpret results and draw reasonable and logical conclusions. In addition, graduates will be able to rationally estimate the solution to a problem, apply appropriate techniques to arrive at a solution, test the correctness of the solution, and interpret their results.


## Performance/Skills-Based

All our graduates will be able to:

- Employ critical thinking and the scientific method to design, carry out, record, analyze and communicate the results of chemical/biochemical experiments. This includes the ability to identify or create an appropriate model, formulate a hypothesis, choose an appropriate set of tools and techniques, and design an experiment that tests the hypothesis and analyze the results from that experiment drawing sound scientific conclusions from the results obtained. In this context, they must be able to locate, identify and critically evaluate the chemical/biochemical literature.
- Develop the interpersonal skills to function cooperatively in a team setting.
- Handle, synthesize, purify, and characterize new and existing substances. This includes knowing the proper procedures and regulations for the safe handling, use and disposal of chemicals.
- Be proficient in the use of both classical and modern tools (e.g., instrumentation, techniques, software) for analysis of chemical systems. Demonstrate effective scientific communication skills, both orally and in writing, to a range of audience levels and for a variety of purposes.
- Understand how scientific information is shared between peers in modern science, including responsible conduct for acknowledging prior and current contributions.


## Affective

All our graduates will be able to:

- Demonstrate an awareness of the benefits and impacts of chemistry related to the environment, society, and other disciplines outside the scientific community. Be prepared to contribute solutions to society's challenges at the intersection of science and society.
- Successfully pursue their career objectives in advanced education in professional and/or graduate schools, in a scientific career in government or industry, in a teaching career in the school systems, or in a related career following graduation.
- Understand and apply ethics and values to all professional activities.


## General Education offerings

Department of Chemistry and Biochemistry
Our general education offerings are designed to enable individuals to form a solid foundation in the conceptual and quantitative thinking that underlies the theories and models that form the basis for reasoning about chemical phenomena.

By introducing and developing a broad set of chemical concepts students will demonstrate an understanding of structure, chemical properties, and reactions of chemicals and biomolecules. Students learn to connect this theoretical understanding to the experimental methods used to test those theories and models. In addition, students will be able to rationally estimate the solution to a problem, apply appropriate techniques to arrive at a solution, test the correctness of the solution and interpret their results.

Our curriculum is infused with strategies and opportunities that enable students to understand how scientific information benefits and impacts society, the environment, and other disciplines outside the scientific community. As students practice critical thinking and become acquainted with the scientific method to analyze and communicate the results of chemical/biochemical experiments, they begin to understand and apply ethics in a multifaceted context of knowledge creation and the impact of that knowledge on society. Our students are prepared to contribute solutions to society's challenges at the intersection of science and society.

Our courses support the development of interpersonal skills to function cooperatively in a team setting and enable students to develop effective scientific communication skills, both orally and in writing to a range of audience levels and for a variety of purposes.

## Curriculum map

Department of Chemistry and Biochemistry/Biochemistry Major
Learning outcomes (LOs): Having completed a major in Biochemistry, a student will be able to:

1. Master a broad set of chemical concepts concerning the fundamentals in the basic areas of the discipline (organic, inorganic, analytical, physical and biochemistry). Students will demonstrate an understanding of structure, chemical properties, and reactions of chemicals and biomolecules.
2. Demonstrate a firm foundation in the conceptual, quantitative, and computational thinking that underlies the theories and models that form the basis for reasoning about molecular systems. Students should be able to connect this theoretical understanding to the experimental methods used to test those theories and models.
3. Demonstrate excellent critical thinking and problem solving abilities. S/he will be able to integrate chemical concepts and ideas learned in lecture courses with skills learned in laboratories to formulate hypotheses, propose and perform experiments, collect data, compile and interpret results and draw reasonable and logical conclusions. In addition, graduates will be able to rationally estimate the solution to a problem, apply appropriate techniques to arrive at a solution, test the correctness of the solution, and interpret their results.
4. Employ critical thinking and the scientific method to design, carry out, record, analyze and communicate the results of chemical/biochemical experiments. This includes the ability to identify or create an appropriate model, formulate a hypothesis, choose an appropriate set of tools and techniques, and design an experiment that tests the hypothesis and analyze the results from that experiment drawing sound scientific conclusions from the results obtained...
5. Develop the interpersonal skills to function cooperatively in a team setting.
6. Handle, synthesize, purify, and characterize new and existing substances. This includes knowing the proper procedures and regulations for the safe handling, use and disposal of chemicals.
7. Be proficient in the use of both classical and modern tools (e.g., instrumentation, techniques, software) for analysis of chemical systems. Demonstrate effective scientific communication skills, both orally and in writing, to a range of audience levels and for a variety of purposes.
8. Understand how scientific information is shared between peers in modern science, including responsible conduct for acknowledging prior and current contributions.
9. Demonstrate an awareness of the benefits and impacts of chemistry related to the environment, society, and other disciplines outside the scientific community. Be prepared to contribute solutions to society's challenges at the intersection of science and society.
10. Successfully pursue their career objectives in advanced education in professional and/or graduate schools, in a scientific career in government or industry, in a teaching career in the school systems, or in a related career following graduation.
11. Understand and apply ethics and values to all professional activities

Library-In this context, they must be able to locate, identify and critically evaluate the chemical/biochemical literature

Key: I = introduces outcome; D = develops outcome; $\mathrm{A}=$ assesses mastery of outcome


|  |  | Foundational Knowledge/Theory |  |  |  |  |  |  |  |  | Performance/Skills Based |  |  |  |  |  | Affective |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Course(s) | Title/description | $\begin{gathered} \text { LO } \\ 1 \\ \hline \end{gathered}$ | Analytical | Biochemistry | Inorganic | Organic | Physical | Related Fields | $\begin{gathered} \mathrm{LO} \\ 2 \end{gathered}$ | $\begin{gathered} \mathrm{LO} \\ 3 \end{gathered}$ | $\begin{gathered} \mathrm{LO} \\ 4 \end{gathered}$ | $\begin{gathered} \hline 10 \\ 5 \end{gathered}$ | $\begin{gathered} 10 \\ 6 \end{gathered}$ | $10$ | $\begin{gathered} \hline 10 \\ 8 \end{gathered}$ | Library | ${ }_{9}^{10}$ | LO 10 | LO 11 |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| CH410 | Electrochemistry (lecturenab) | DA | A |  | D | D | A |  | A | A | A | D |  | A | D | A | D |  | D |
| CH 410 | Biochemistry of Disease (lecture) | ID |  | ID |  |  |  |  | ID | IDA |  |  |  | DA | D | DA | ID |  |  |
| CH 410 | Structural Bicchemistry (lecture) | DA |  | DA |  |  |  |  | DA | IDA |  |  |  | DA | D | DA | 10 |  |  |
| CH 410 | Cytoskeleton (lecture) | DA |  | DA |  |  |  |  | DA | IDA |  |  |  | DA | D | DA | 10 |  |  |
| CH410 | Biophysics \& Evolution (lecture) | DA |  | DA |  |  |  |  | DA | IDA |  |  |  | DA. | D | DA | 1 D |  |  |
| CH 420 | Physical Organic Chemistry I (lecture) | A |  |  |  | A |  |  | A | A |  |  |  |  |  | A |  |  | D |
| CH 421 | Physical Organic Chemistry \|| (lecture) | A |  |  |  | A |  |  | A | A |  |  |  |  |  | A |  |  | D |
| CH 429 | Instrumental Analvsis Laboratory | DA | DA |  | A |  | A | A | DA | DA | DA |  | D | DA | DA | A | D | D | D |
| CH 432 | Inorganic Chemistry: Bioionorganic Chernistry (lecture) | DA |  | D |  | A | D |  | A | A | A |  |  |  | A | A | A |  |  |
| CH 433 | Inorganic Chemistry: Solid-state Chernistry (lecture) | A |  |  | A |  | 0 |  | A | A | A |  |  | A | A | A | A |  | A |
| CH 437 | Inorganic Chemistry Laboratory | A |  |  | A |  |  |  | A | A | A | A | A | A | D | A | D | D | D |
| CH 441 | Quantum Chemistry (lecture) | DA |  |  |  |  | DA |  | DA | DA | DA | DA |  |  |  | DA | DA | DA | DA |
| CH 442 | Quantum Chemistry (lecture) | DA |  |  |  |  | DA |  | DA | DA | DA | DA |  |  |  | DA | DA | DA | DA |
| CH 443 | Quantum Chemistry and Spectroscopy (lecture) | DA |  |  |  |  | DA |  | DA | DA | DA | DA |  |  |  | DA | DA | DA | DA |
| CH 444 | Chemical Thermodynamics (lecture) | DA |  |  |  |  | DA |  | DA | DA | DA | DA |  |  |  | DA | DA | DA | DA |
| CH445 | Statistical Mechanics (lecture) | DA |  |  |  |  | DA | DA | DA | DA | DA | D |  |  |  | D | D |  |  |
| C. 447 | Computational Chemistry (lecturenab) | ${ }^{\text {DA }}$ |  | 0 | D | D | DA |  | DA | DA | DA |  |  |  |  | DA |  |  |  |
| $\mathrm{CH}^{4} 452$ | Advanced Oraanic Chemistry Stereochemistiv and Reactions (lecture) | A |  |  | D | D | IDA |  | A |  | 10A |  |  |  |  | DA |  | D | D |
| CH464 | RNA Biochemistry (lecture) | DA |  | DA |  |  |  |  | 10 | DA |  |  |  | DA | D | DA | 10 |  |  |
| Advanced Electives |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| B1322 | Cell ioplogr (lecture) | 10 |  |  |  |  |  | 10 |  | 10 |  |  |  |  |  |  |  |  |  |
|  | Develoomental Biology (lecture) | ID |  |  |  |  |  | 10 |  | 10 |  |  |  |  |  |  |  |  |  |
| B1360 | Neurobioloay (lecture) | 10 |  |  |  |  |  | 10 |  | 10 |  |  |  |  |  |  |  |  |  |
| 81.423 | Protein toxins in cell biolocy (lecture) | DA |  |  |  |  |  | DA |  | 10 |  |  |  |  |  |  |  |  |  |
| B1 424 | Advanced Molecular Genetics (lecture) | DA |  |  |  |  |  | DA |  | 10 |  |  |  |  |  |  |  |  |  |
| B1 426 | Genetics of Cancer (lecture) | DA |  |  |  |  |  | DA |  | 10 |  |  |  |  |  |  |  |  |  |
| B1/ 428 | Developmental Genetics (lecture) | DA |  |  |  |  |  | DA |  | 10 |  |  |  |  |  |  |  |  |  |
| P1433 | Bacterat-Host interactions (ilecture) | IDA |  |  |  |  |  | IDA |  | 10 |  |  |  |  |  |  |  |  |  |
| B1461 | Systems Neuroscience (lecture) | IDA |  |  |  |  |  | IDA |  | ID |  |  |  |  |  |  |  |  |  |
| 81465 | Developmentail Nesurobiologocur (lecture) | IDA |  |  |  |  |  | 1 IDA |  | 10 |  |  |  |  |  |  |  |  |  |
| B1480 | Evolution of Development (lecture) |  |  |  |  |  |  | 1 DA |  | 10 |  |  |  |  |  |  |  |  |  |
| B1484 | Molecular Evolution (lecture) |  |  |  |  |  |  | DA |  | 18 |  |  |  |  |  |  |  |  |  |
| 81493 | Genomic Approaches and Analysis (lecture) |  |  |  |  |  |  | IDA |  | ID |  |  |  |  |  |  |  |  |  |

## Learning outcomes explanations

Department of Chemistry and Biochemistry/Biochemistry Major
Majors in chemistry and biochemistry provide training for students planning careers in the chemical and biological sciences and also for those in biology, health related disciplines, earth sciences, secondary education, business, journalism and
law. Approximately one quarter of the UO undergraduate population will take a course in the Department of Chemistry and Biochemistry. The Department's curriculum is designed to satisfy the diverse needs of all these students.

Chemistry and biochemistry graduates complete an integrated, rigorous program that includes foundational course work in chemistry and biochemistry and additional course work in related fields. The ACS-certified degree further emphasizes laboratory experience and the development of professional skills. Undergraduate research and other educational activities outside the traditional classroom are essential components of these majors. Undergraduate majors also benefit from taking graduate courses in synthetic, physical, materials, computational chemistry, biochemistry molecular biology and modern instrumental techniques.

Graduates of our program will have a robust set of fundamental competencies that are knowledge-based, performance/skills-based, and affective.

## Foundational knowledge/theory

All our graduates will be able to:

- Master a broad set of chemical concepts concerning the fundamentals in the basic areas of the discipline (organic, inorganic, analytical, physical and biochemistry). Students will demonstrate an understanding of structure, chemical properties, and reactions of chemicals and biomolecules.
- Demonstrate a firm foundation in the conceptual, quantitative, and computational thinking that underlies the theories and models that form the basis for reasoning about molecular systems. Students should be able to connect this theoretical understanding to the experimental methods used to test those theories and models.
- Demonstrate excellent critical thinking and problem solving abilities. S/he will be able to integrate chemical concepts and ideas learned in lecture courses with skills learned in laboratories to formulate hypotheses, propose and perform experiments, collect data, compile and interpret results and draw reasonable and logical conclusions. In addition, graduates will be able to rationally estimate the solution to a problem, apply appropriate techniques to arrive at a solution, test the correctness of the solution, and interpret their results.


## Performance/Skills-Based

All our graduates will be able to:

- Employ critical thinking and the scientific method to design, carry out, record, analyze and communicate the results of chemical/biochemical experiments. This includes the ability to identify or create an appropriate model, formulate a hypothesis, choose an appropriate set of tools and techniques, and design an experiment that tests the hypothesis and analyze the results from that experiment drawing sound scientific conclusions from the results obtained. In this context, they must be able to locate, identify and critically evaluate the chemical/biochemical literature.
- Develop the interpersonal skills to function cooperatively in a team setting.
- Handle, synthesize, purify, and characterize new and existing substances. This includes knowing the proper procedures and regulations for the safe handling, use and disposal of chemicals.
- Be proficient in the use of both classical and modern tools (e.g., instrumentation, techniques, software) for analysis of chemical systems. Demonstrate effective scientific communication skills, both orally and in writing, to a range of audience levels and for a variety of purposes.
- Understand how scientific information is shared between peers in modern science, including responsible conduct for acknowledging prior and current contributions.


## Affective

All our graduates will be able to:

- Demonstrate an awareness of the benefits and impacts of chemistry related to the environment, society, and other disciplines outside the scientific community. Be prepared to contribute solutions to society's challenges at the intersection of science and society.
- Successfully pursue their career objectives in advanced education in professional and/or graduate schools, in a scientific career in government or industry, in a teaching career in the school systems, or in a related career following graduation.
- Understand and apply ethics and values to all professional activities.


## General Education offerings

Department of Chemistry and Biochemistry

Our general education offerings are designed to enable individuals to form a solid foundation in the conceptual and quantitative thinking that underlies the theories and models that form the basis for reasoning about chemical phenomena.

By introducing and developing a broad set of chemical concepts students will demonstrate an understanding of structure, chemical properties, and reactions of chemicals and biomolecules. Students learn to connect this theoretical understanding to the experimental methods used to test those theories and models. In addition, students will be able to rationally estimate the solution to a problem, apply appropriate techniques to arrive at a solution, test the correctness of the solution and interpret their results.

Our curriculum is infused with strategies and opportunities that enable students to understand how scientific information benefits and impacts society, the environment, and other disciplines outside the scientific community. As students practice critical thinking and become acquainted with the scientific method to analyze and communicate the results of chemical/biochemical experiments, they begin to understand and apply ethics in a multifaceted context of knowledge creation and the impact of that knowledge on society. Our students are prepared to contribute solutions to society's challenges at the intersection of science and society.

Our courses support the development of interpersonal skills to function cooperatively in a team setting and enable students to develop effective scientific communication skills, both orally and in writing to a range of audience levels and for a variety of purposes.

